AQA

ASSESSMENT and
QUALIFICATIONS
ALLIANCE

General Certificate of Education

Physics 5451
Specification A

PA02 Mechanics and Molecular Kinetic Theory

Mark Scheme

2006 examination – June series

Mark schemes are prepared by the Principal Examiner and considered, together with the relevant questions, by a panel of subject teachers. This mark scheme includes any amendments made at the standardisation meeting attended by all examiners and is the scheme which was used by them in this examination. The standardisation meeting ensures that the mark scheme covers the candidates' responses to questions and that every examiner understands and applies it in the same correct way. As preparation for the standardisation meeting each examiner analyses a number of candidates' scripts: alternative answers not already covered by the mark scheme are discussed at the meeting and legislated for. If, after this meeting, examiners encounter unusual answers which have not been discussed at the meeting they are required to refer these to the Principal Examiner.

It must be stressed that a mark scheme is a working document, in many cases further developed and expanded on the basis of candidates' reactions to a particular paper. Assumptions about future mark schemes on the basis of one year's document should be avoided; whilst the guiding principles of assessment remain constant, details will change, depending on the content of a particular examination paper.

Instructions to Examiners

1   Give due credit to alternative treatments which are correct. Give marks for what is correct; do not deduct marks because the attempt falls short of some ideal answer. Where marks are to be deducted for particular errors specific instructions are given in the marking scheme.

2   Do not deduct marks for poor written communication. Refer the script to the Awards meeting if poor presentation forbids a proper assessment. In each paper candidates may be awarded up to two marks for the Quality of Written Communication in cases of required explanation or description. Use the following criteria to award marks:

  2 marks: Candidates write legibly with accurate spelling, grammar and punctuation; the answer containing information that bears some relevance to the question and being organised clearly and coherently. The vocabulary should be appropriate to the topic being examined.

  1 mark: Candidates write with reasonably accurate spelling, grammar and punctuation; the answer containing some information that bears some relevance to the question and being reasonably well organised. Some of the vocabulary should be appropriate to the topic being examined.

  0 marks: Candidates who fail to reach the threshold for the award of one mark.

3   An arithmetical error in an answer should be marked AE thus causing the candidate to lose one mark. The candidate's incorrect value should be carried through all subsequent calculations for the question and, if there are no subsequent errors, the candidate can score all remaining marks (indicated by ticks). These subsequent ticks should be marked CE (consequential error).

4   With regard to incorrect use of significant figures, normally two, three or four significant figures will be acceptable. Exceptions to this rule occur if the data in the question is given to, for example, five significant figures as in values of wavelength or frequency in questions dealing with the Doppler effect, or in atomic data. In these cases up to two further significant figures will be acceptable. The maximum penalty for an error in significant figures is **one mark per paper**. When the penalty is imposed, indicate the error in the script by SF and, in addition, write SF opposite the mark for that question on the front cover of the paper to obviate imposing the penalty more than once per paper.

5   No penalties should be imposed for incorrect or omitted units at intermediate stages in a calculation or which are contained in brackets in the marking scheme. Penalties for unit errors (incorrect or omitted units) are imposed only at the stage when the final answer to a calculation is considered. The maximum penalty is **one mark per question**.

6   All other procedures, including the entering of marks, transferring marks to the front cover and referrals of scripts (other than those mentioned above) will be clarified at the standardising meeting of examiners.
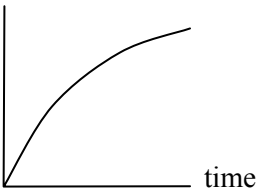
PA02: Mechanics and Molecular Kinetic Theory

Question 1			
(a)		momentum ✓ kinetic energy ✓	**2**
(b)	(i)	$450\,\mathrm{m\,s^{-1}}$ ✓ in the opposite direction ✓	**4**
	(ii)	$\Delta p = 8.0 \times 10^{-26} \times 900$ ✓ $\quad = 7.2 \times 10^{-23}\,\mathrm{N\,s}$ ✓	
(c)		force is exerted on molecule by wall ✓ to change its momentum ✓ molecule must exert an equal but opposite force on wall ✓ in accordance with Newton's second or third law ✓	**4**
		Total	**10**

Question 2			
(a)		potential energy to kinetic energy ✓ mention of thermal energy and friction ✓	**2**
(b)		(use of $\frac{1}{2}mv^2 = mgh$ gives) $\frac{1}{2}v_h{}^2 = 9.81 \times 1.5$ ✓ $v_h = 5.4(2)\,\mathrm{m\,s^{-1}}$ ✓ (assumption) energy converted to thermal energy is negligible ✓	**3**
(c)		component of weight down the slope causes acceleration ✓ this component decreases as skateboard moves further down the slope ✓ air resistance/friction increases (with speed) ✓	**max 2**
(d)	(i)	distance $(= 0.42 \times 5.4) = 2.3\,\mathrm{m}$ ✓ $(2.27\,\mathrm{m})$ (allow C.E. for value of v_h from (b))	**5**
	(ii)	$v_v = 9.8 \times 0.42$ ✓ $\quad = 4.1(1)\,\mathrm{m\,s^{-1}}$ ✓	
	(iii)	$v^2 = 4.1^2 + 5.4^2$ ✓ $v = 6.8\,\mathrm{m\,s^{-1}}$ ✓ $(6.78\,\mathrm{m\,s^{-1}})$ (allow C.E. for value of v_h from (b))	
		Total	**12**

Question 3			
(a)		resultant force zero ✓ resultant torque about any point zero ✓	**2**
(b)	(i)	force due to wire P $= 5.0 - 2.0 = 3.0\,\mathrm{N}$ ✓	**3**
	(ii)	(moments give) $5.0 \times d = 2.0 \times 0.90$ ✓ $d = 0.36\,\mathrm{m}$ ✓	
		Total	**5**

Question 4		
(a)	(use of $\Delta Q = mcT$ gives) $\Delta Q_1 = 1.5 \times 4200 \times 18$ ✔ $\qquad = 1.134 \times 10^5$ (J) ✔ $\Delta Q_2 = 1.5 \times 3.4 \times 10^5 = 5.1 \times 10^5$ (J) ✔ total energy released ($= 1.134 \times 10^5 + 5.1 \times 10^5$) $\qquad\qquad\qquad\qquad = 6.2 \times 10^5$ J ✔ (6.23×10^5 J)	**4**
(b)	(ice) requires energy to melt [or mention of latent heat] ✔ stays at 0°C (for longer) (or cools for longer) ✔ (or extracts more energy from the drink)	**2**
	Total	**6**

Question 5			
(a)	(i)	(use of $pV = nRT$ gives) $n = \dfrac{1.0 \times 10^5 \times 5.0 \times 10^{-3}}{8.31 \times 310}$ ✔ $\qquad\qquad = 0.194$ moles ✔	**4**
	(ii)	average kinetic energy $= {}^3\!/_2 \times 1.38 \times 10^{-23} \times 310$ ✔ $\qquad\qquad = 6.4 \times 10^{-21}$ J ✔	
(b)		same temperature ✔ (hence) same **average** kinetic energy ✔ (since) kinetic energy is $\frac{1}{2}mv^2$ ✔ nitrogen **molecules** must have a higher mean square speed ✔	**max 3**
		Total	**7**

Question 6			
(a)	(i)	(use of $a = \dfrac{\Delta v}{\Delta t}$ gives) $a = \dfrac{4.5}{3600}$ ✔ $\qquad\qquad\qquad = 1.25 \times 10^{-3}\,\text{m s}^{-2}$ ✔	**4**
	(ii)	(use of $v^2 = u^2 + 2as$ gives) $0 = 4.5^2 - 2 \times 1.25 \times 10^{-3} \times s$ ✔ $s\left(= \dfrac{20.25}{2.5 \times 10^{-3}}\right) = 8.1 \times 10^3\,\text{m}$ ✔	
(b)		increasing curve ✔ correct curve ✔	**2**
(c)		gradient (slope) of graph represents speed ✔ hence graph has decreasing gradient ✔	**2**
		Total	**8**

Quality of Written Communication: Q1 (c) and Q5 (c)	**2**